

Supplementary material to Exploring the impact of Al-based electrolytes on the charge storage behaviour of vine shoots derived carbon

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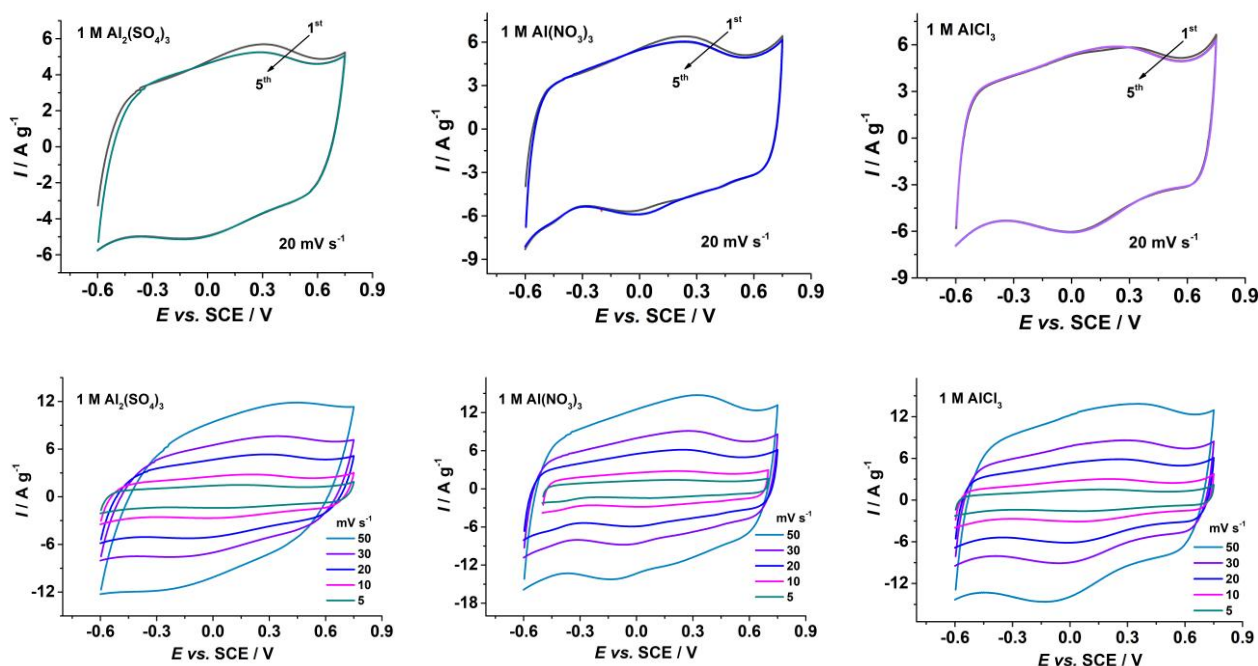


Figure S1. Cyclic voltammograms of AC_{vs}700 in 1 M Al₂(SO₄)₃, Al(NO₃)₃ and AlCl₃ at a common scan rate 20 mV s⁻¹ and at different scan rates

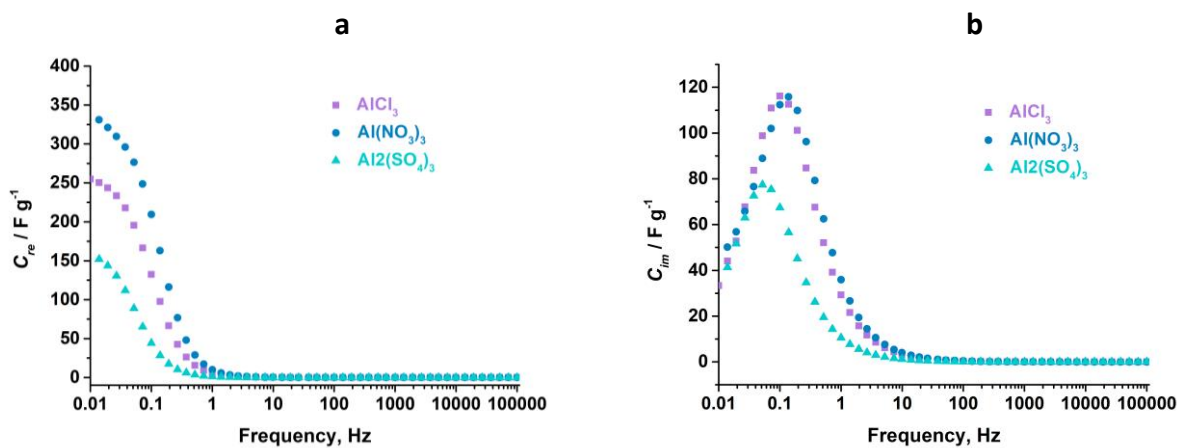


Figure S2. a -real and b - imaginary part of the capacitance as the function of the frequency plot of ACvs700 pristine electrode in 1 M aqueous solutions of $Al_2(SO_4)_3$, $Al(NO_3)_3$ and $AlCl_3$ measured in a three-electrode cell

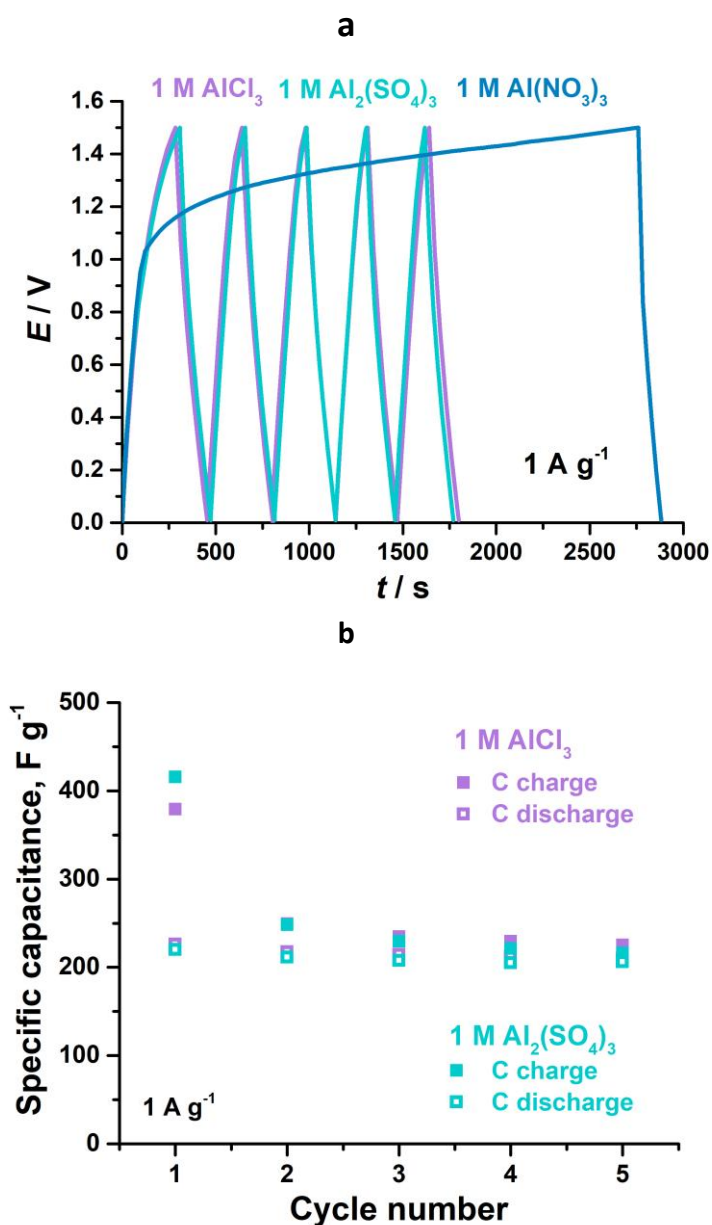


Figure S3. a - charge/discharge curves of ACvs700 cells in different Al-based electrolytes and b - the specific capacitance of ACvs700 electrode in $Al_2(SO_4)_3$ and $AlCl_3$

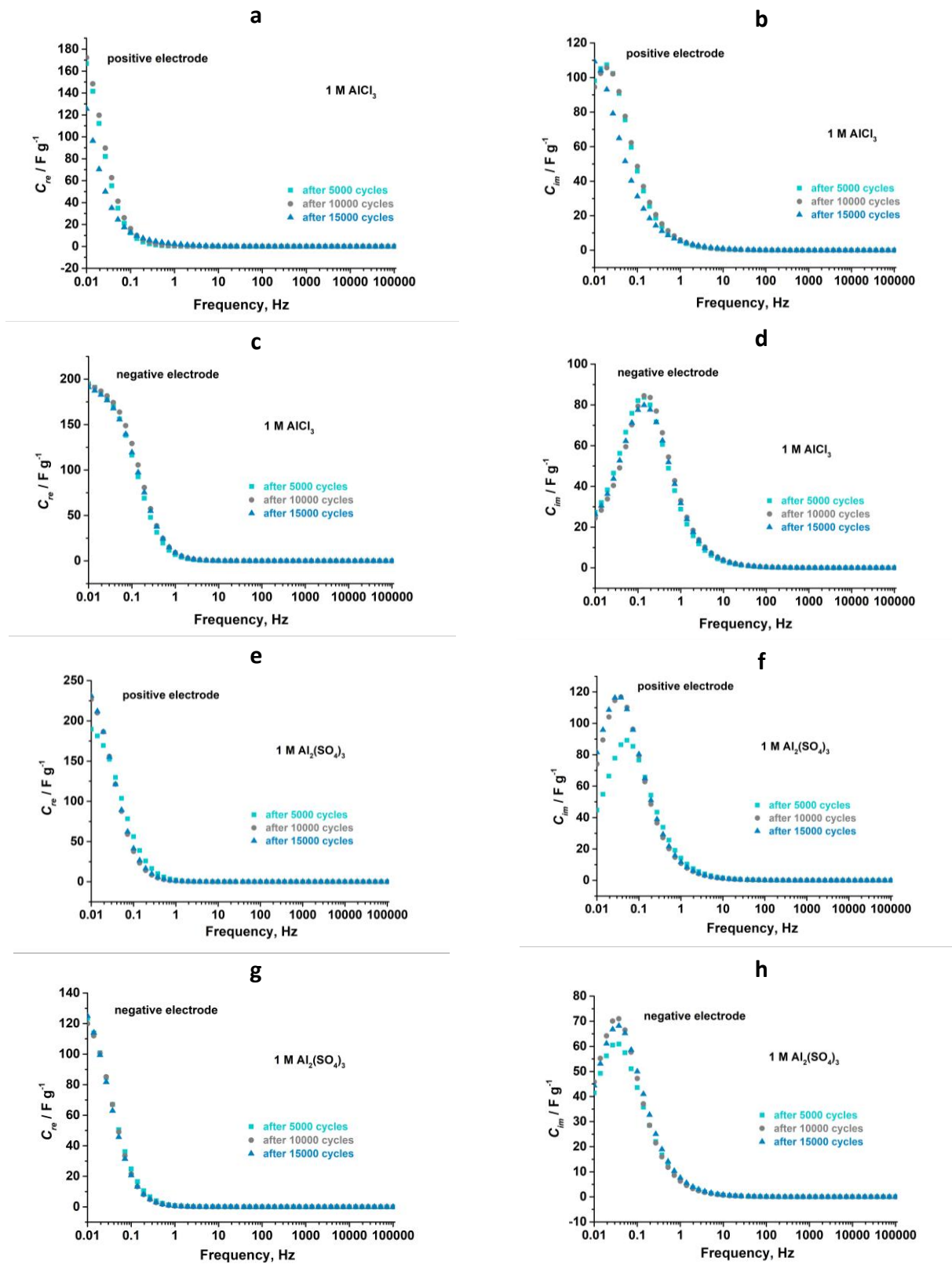


Figure S4. Real and imaginary part of the capacitance as the function of the frequency plot of AC_{vs700} positive and negative electrode after 5,000, 10,000 and 15,000 cycles in 1 M aqueous solutions of $AlCl_3$ (a, b, c, d) and $Al_2(SO_4)_3$ (e, f, g, h), measured in a three-electrode cell